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Multi-temperature hydrodynamic equations from kinetic theory for rarefied gas mixtures

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Summary

Preliminaries: physical setting and some bibliography

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Rescaled kinetic equations for polyatomic gases and derivation of multi-temperature Euler closure

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 Explicit exchange rates for simple collision models and analysis of the hydrodynamic system

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Mixture of Q rarefied gases A_s $s=1,\ldots,Q$

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set of integro-differential equations of Boltzmann type for the distribution functions $f_s(\mathbf{x}, \mathbf{v}, t)$, $s = 1, \dots, Q$

$$\frac{\partial f_s}{\partial t} + \mathbf{v} \cdot \nabla_{\mathbf{x}} f_s = \sum_{r=1}^{Q} I_{sr}[f_s, f_r]$$

 $I_{sr}[f_s, f_r]$: Boltzmann collision operator describing the effects of binary collisions involving particles of gases A_s and A_r

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- Elastic collisions preserve number densities of single species, global momentum and global kinetic energy
- More complex situations like polyatomic gases or chemically reacting mixtures may involve exchange of internal or chemical energy (with conservation of total energy)

Some bibliography on the kinetic approach

Kinetic models of Boltzmann type, possibly accounting for internal energy levels and/or dissociation and recombination processes

[Chapman, Cowling (1970), Groppi, Spiga (1999), Giovangigli (1999), Groppi, Rossani, Spiga (2000), Polewczak (2001), Desvillettes, Monaco, Salvarani (2005), ...]

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Kinetic models of BGK type including chemical reactions [Andries, Aoki, Perthame (2002), Monaco, Pandolfi Bianchi (2004), Groppi, Spiga (2004), Kremer, Pandolfi Bianchi, Soares (2006), Asinari (2008), Bisi, Cáceres, Spiga (2010), ...]

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- Hydrodynamic closures in elastically dominated regimes [Review: Bisi, Groppi, Spiga (2005), ...]

Mixtures diffusing in a background medium: hydrodynamic equations of reaction—diffusion type [De Masi, Ferrari, Lebowitz (1986), De Masi, Presutti (1991), Spigler, Zanette (1993), Bisi, Desvillettes (2006), Bisi, Spiga (2006), ...] Mixtures diffusing in a background medium: hydrodynamic equations of reaction—diffusion type [De Masi, Ferrari, Lebowitz (1986), De Masi, Presutti (1991), Spigler, Zanette (1993), Bisi, Desvillettes (2006), Bisi, Spiga (2006), ...]

Remark

These strategies usually lead to fluid–dynamic descriptions involving number densities of single species N_s , global mass velocity of the mixture \mathbf{u} , global temperature T

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Aim of our work

We present a formal derivation, starting from suitably rescaled kinetic equations, of an hydrodynamic description involving temperatures and velocities of single gases

In the description of several physical problems, like plasmas, aerothermodynamics, thermally non-equilibrium conditions [Park (1990), Bose (2003), Kustova, Nagnibeda (2006), ...]

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Here hydrodynamic variables are mass densities, global velocity, and a unique translational temperature for the mixture plus an internal (vibrational) temperature for each species

Kinetic model for polyatomic gases

[Groppi, Spiga (1999), Desvillettes, Monaco, Salvarani (2005)]

• Each species A_s , $s=1,\ldots,Q$, is endowed with a structure of N>1 discrete energy levels

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- The QN different components are ordered in such a way that the s-th gas may be regarded as the equivalence class of the indices $i \equiv s$ modulo Q
- If A_i , $1 \le i \le QN$, denotes the general component, and E_i the corresponding energy of its state, the general binary interaction is written as

$$A_i + A_j \rightleftharpoons A_h + A_k \qquad i \equiv h \qquad j \equiv k$$

The net increase of internal energy $\Delta E_{ij}^{hk} = E_h + E_k - E_i - E_j$ must be compensated by an opposite variation of the kinetic energies

Kinetic equations for functions $f_i(\mathbf{x}, \mathbf{v}, t)$, $i = 1, \dots, NQ$

$$\frac{\partial f_i}{\partial t} + \mathbf{v} \cdot \nabla_{\mathbf{x}} f_i = \sum_{(j,h,k) \in D_i} \iint K_i^{ijhk} [\underline{f}](\mathbf{v}, \mathbf{w}, \hat{\mathbf{n}}') d\mathbf{w} d\hat{\mathbf{n}}',$$

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$$K_{i}^{ijhk} [\underline{f}](\mathbf{v}, \mathbf{w}, \hat{\mathbf{n}}') = \Theta(g^{2} - \delta_{ij}^{hk}) B_{ij}^{hk} (g, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}')$$

$$\cdot \left[\left(\frac{\mu_{ij}}{\mu_{hk}} \right)^{3} f_{h} \left(\mathbf{v}_{ij}^{hk} \right) f_{k} \left(\mathbf{w}_{ij}^{hk} \right) - f_{i}(\mathbf{v}) f_{j}(\mathbf{w}) \right]$$

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with

- $\mathbf{g} = \mathbf{v} \mathbf{w} = g \,\hat{\mathbf{n}}$ relative velocity
- B_{ij}^{hk} collision kernel (relative speed times cross section)
- μ_{ij} reduced mass
- Θ unit step function introduces a threshold for the collision if $\delta^{hk}_{ij}=2\,\Delta\!E^{hk}_{ij}/\mu_{ij}>0$
- \mathbf{v}_{ij}^{hk} , \mathbf{w}_{ij}^{hk} post–collision velocities

Collision equilibria

$$\mathcal{M}_i(\mathbf{v}) = n_i \left(\frac{m_s}{2\pi KT}\right)^{3/2} \exp\left[-\frac{m_s}{2KT}(\mathbf{v} - \mathbf{u})^2\right] \ \forall i \equiv s, \ \forall s = 1, \dots, Q$$

where

$$n_i = \psi_i(E_i, T) N_s \qquad \psi_i(E_i, T) = \frac{\exp\left(-\frac{E_i - E_s}{KT}\right)}{\sum_{i \equiv s} \exp\left(-\frac{E_i - E_s}{KT}\right)} = \frac{\exp\left(-\frac{E_i - E_s}{KT}\right)}{Z_s(T)}$$

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–functional

$$H = \sum_{i=1}^{Q} \sum_{i=n} \int f_i \log f_i \, d\mathbf{v}$$

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Q+4 independent **Conservation laws**:

 $N_s = \sum_{i \equiv s} n_i$, $s = 1, \dots, Q$, number densities of single species

u global momentum

 $\frac{3}{2}NKT + \sum_{i=1}^{QN} E_i n_i$ total (kinetic + internal) energy

Rescaled kinetic model

Among all possible interactions

$$A_i + A_j \rightleftharpoons A_h + A_k \qquad i \equiv h \qquad j \equiv k$$

we assume the mean free path for collisions between components of the same species much shorter than for collisions between components of different species:

FAST:
$$i \equiv j \equiv h \equiv k$$

SLOW:
$$i \not\equiv j$$
 $i \equiv h$ $j \equiv k$

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Scaled kinetic equations

$$\frac{\partial f_i}{\partial t} + \mathbf{v} \cdot \nabla_{\mathbf{x}} f_i = \frac{1}{\epsilon} I_i^{\mathbf{F}\mathbf{A}} + I_i^{\mathbf{S}\mathbf{L}}$$

Investigation of the leading operators $I_i^{\it FA}$

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Dominant collision equilibria

$$f_i^M(\mathbf{v}) = \frac{N_s}{Z_s(T_s)} \left(\frac{m_s}{2\pi K T_s}\right)^{3/2} \exp\left[-\frac{m_s}{2K T_s} (\mathbf{v} - \mathbf{u}_s)^2 - \frac{E_i - E_s}{K T_s}\right]$$

$$\forall i \equiv s, \quad \forall s = 1, \dots, Q,$$

with free parameters N_s , \mathbf{u}_s , T_s

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 \Rightarrow 5Q collision invariants corresponding to preservation of number density, momentum, and kinetic energy within each species

 \Rightarrow 5Q macroscopic "conservation" (for the fast operator) equations, that we aim at closing at Euler accuracy

$$\frac{\partial N_s}{\partial t} + \nabla_{\mathbf{x}} \cdot (N_s \mathbf{u}_s) = 0$$

$$\frac{\partial}{\partial t} (\rho_s \mathbf{u}_s) + \nabla_{\mathbf{x}} \cdot \sum_{i \equiv s} (\rho_i \mathbf{u}_i \otimes \mathbf{u}_i + \mathbf{P}_i) = \mathbf{R}_s$$

$$\frac{\partial}{\partial t} \left[\sum_{i \equiv s} \left(\frac{1}{2} \rho_i u_i^2 + \frac{3}{2} n_i K T_i + E_i n_i \right) \right] + \nabla_{\mathbf{x}} \cdot \left\{ \sum_{i \equiv s} \left[\left(\frac{1}{2} \rho_i u_i^2 + \frac{3}{2} n_i K T_i + E_i n_i \right) \right] + \nabla_{\mathbf{x}} \cdot \left\{ \sum_{i \equiv s} \left[\left(\frac{1}{2} \rho_i u_i^2 + \frac{3}{2} n_i K T_i + E_i n_i \right) \right] \right\} = S_s$$

where \mathbb{R}_s and S_s are collision contributions due to the slow interactions

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They are provided by weak forms of the Boltzmann operators corresponding to the test functions $\varphi_i(\mathbf{v}) = m_i \mathbf{v}$ and

$$\varphi_i(\mathbf{v}) = \frac{1}{2} \, m_i v^2 + E_i$$

$$\mathbf{R}_{s} = \sum_{r \neq s} \sum_{i,h \equiv s} \int \int \int \int m_{s} (\mathbf{v}_{ij}^{hk} - \mathbf{v}) B_{ij}^{hk}(g, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}') \Theta(g^{2} - \delta_{ij}^{hk}) f_{i}(\mathbf{v}) f_{j}(\mathbf{w}) d\mathbf{v} d\mathbf{w} d\hat{\mathbf{n}}'$$

$$S_{s} = \sum_{r \neq s} \sum_{i,h \equiv s} \int \int \int \int \left\{ \frac{1}{2} m_{s} \left[(v_{ij}^{hk})^{2} - v^{2} \right] + E_{h} - E_{i} \right\}$$

$$\times B_{ij}^{hk}(g, \hat{\mathbf{n}} \cdot \hat{\mathbf{n}}') \Theta(g^{2} - \delta_{ij}^{hk}) f_{i}(\mathbf{v}) f_{j}(\mathbf{w}) d\mathbf{v} d\mathbf{w} d\hat{\mathbf{n}}'$$

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They correctly reproduce the overall conservations of momentum and energy

$$\sum_{s=1}^{Q} \mathbf{R}_s = \mathbf{0} \qquad \qquad \sum_{s=1}^{Q} S_s = 0$$

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$$\sum_{s=1}^{Q} \mathbf{R}_s = \mathbf{0} \qquad \qquad \sum_{s=1}^{Q} S_s = 0$$

The zero—order closure is achieved by substituting into macroscopic equations the fast collision equilibrium $f_i^M(N_s, \mathbf{u}_s, T_s)$ for the actual distribution functions

Fluid-dynamic Euler equations

$$\frac{\partial N_s}{\partial t} + \nabla_{\mathbf{x}} \cdot (N_s \mathbf{u}_s) = 0$$

$$\frac{\partial}{\partial t} (\rho_s \mathbf{u}_s) + \nabla_{\mathbf{x}} \cdot (\rho_s \mathbf{u}_s \otimes \mathbf{u}_s) + \nabla_{\mathbf{x}} (N_s K T_s) = \hat{\mathbf{R}}_s$$

$$\frac{\partial}{\partial t} \left(\frac{1}{2} \rho_s u_s^2 + \frac{3}{2} N_s K T_s + N_s \bar{E}_s (T_s) \right) +$$

$$+ \nabla_{\mathbf{x}} \cdot \left[\left(\frac{1}{2} \rho_s u_s^2 + \frac{5}{2} N_s K T_s + N_s \bar{E}_s (T_s) \right) \mathbf{u}_s \right] = \hat{S}_s$$

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where

$$\bar{E}_s(T_s) = \frac{1}{Z_s(T_s)} \sum_{i=s} E_i \exp\left(-\frac{E_i - E_s}{KT_s}\right)$$

and $\hat{\mathbf{R}}_s$, \hat{S}_s have become known functions of the 5Q unknown fields N_s , \mathbf{u}_s , T_s

Evaluation of collision contributions:

crucial steps and difficulties

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1) The product of two Maxwellians at different velocities and temperatures may be cast as

$$f_i^M(\mathbf{v})f_j^M(\mathbf{w}) = \frac{N_s}{Z_s(T_s)} \frac{N_r}{Z_r(T_r)} \left(\frac{m_s}{2\pi K T_s}\right)^{\frac{3}{2}} \left(\frac{m_r}{2\pi K T_r}\right)^{\frac{3}{2}} \exp\left(-\frac{E_i - E_s}{K T_s}\right)^{\frac{3}{2}} \left(\frac{m_r}{2\pi K T_r}\right)^{\frac{3}{2}} \exp\left(-\frac{E_i - E_s}{K T_s}\right)^{\frac{3}{2}} \left(\frac{m_r}{2\pi K T_r}\right)^{\frac{3}{2}} \exp\left(-\frac{E_i - E_s}{K T_s}\right)^{\frac{3}{2}} \exp\left(-\frac{E_i - E_s}{K T_s}\right)$$

where

$$\mathbf{G}_{sr} = \frac{m_s}{m_s + m_r} \mathbf{v} + \frac{m_r}{m_s + m_r} \mathbf{w} \qquad \mathbf{g} = \mathbf{v} - \mathbf{w}$$

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$$\alpha_{sr} = \frac{m_s}{2KT_s} + \frac{m_r}{2KT_r} \qquad \beta_{sr} = \left(\frac{2KT_s}{m_s} + \frac{2KT_r}{m_r}\right)^{-1}$$

$$\gamma_{sr} = \frac{\mu_{sr}}{\alpha_{sr}} \left(\frac{1}{2KT_s} - \frac{1}{2KT_r}\right) \qquad \boldsymbol{\delta}_{sr} = \frac{1}{\alpha_{sr}} \left(\frac{m_s}{2KT_s} \mathbf{u}_s + \frac{m_r}{2KT_r} \mathbf{u}_r\right)$$

2) Angular integrations may be cast in terms of suitably averages of the collision kernel:

$$\bar{B}_{ij}^{hk}(g) = B_{ij}^{hk(0)}(g) - \left(1 - \frac{\delta_{ij}^{hk}}{g^2}\right)^{\frac{1}{2}} B_{ij}^{hk(1)}(g).$$

where

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 \Rightarrow This allows to push further analytical manipulations in polar coordinates, leaving only a one–dimensional integral with respect to $g = |\mathbf{g}|$

$$\hat{\mathbf{R}}_{sr} = -\frac{\mu_{sr}}{2\sqrt{\pi}} \frac{1}{\beta_{sr}^{3/2} |\mathbf{u}_s - \mathbf{u}_r|^2} \frac{N_s}{Z_s(T_s)} \frac{N_r}{Z_r(T_r)}$$

$$\times \sum_{i,h \equiv s} \sum_{j,k \equiv r} \exp\left(-\frac{E_i - E_s}{KT_s} - \frac{E_j - E_r}{KT_r}\right) X_{ij}^{hk} (|\mathbf{u}_s - \mathbf{u}_r|, \beta_{sr}) \frac{\mathbf{u}_s - \mathbf{u}_r}{|\mathbf{u}_s - \mathbf{u}_r|}$$

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$$\times \sum_{i,b=s} \sum_{i,b=s} \exp\left(-\frac{E_i - E_s}{KT_s} - \frac{E_j - E_r}{KT_r}\right) Y_{ij}^{hk} \left(|\mathbf{u}_s - \mathbf{u}_r|, \beta_{sr}, \gamma_{sr}\right)$$

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where, setting $\Delta_{sr}=eta_{sr}^{1/2}\,|\mathbf{u}_s-\mathbf{u}_r|$,

$$X_{ij}^{hk}(|\mathbf{u}_s - \mathbf{u}_r|, \beta_{sr}) = \int_0^\infty \Theta\left(x^2 - \beta_{sr}\delta_{ij}^{hk}\right) \bar{B}_{ij}^{hk}(\beta_{sr}^{-1/2}x)$$

$$\times \left\{ (2\Delta_{sr}x - 1) \exp\left[-(x - \Delta_{sr})^2\right] + (2\Delta_{sr}x + 1) \exp\left[-(x + \Delta_{sr})^2\right] \right\} x dx$$

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Exchange rates for simple models

Maxwell molecule frame

If for a given collision $(i,j) \to (h,k)$ we assume B^{hk}_{ij} depending only on $\hat{\mathbf{n}} \cdot \hat{\mathbf{n}}' \Rightarrow \bar{B}^{hk}_{ij}(g) = B^{hk(0)}_{ij}(g) = \kappa^{hk}_{ij} = \mathrm{constant}$

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(with \mathcal{F}_1 , \mathcal{F}_2 , \mathcal{F}_3 suitable combinations of exponentials and error functions)

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Remark: Unfortunately, this does not make explicit the whole exchange rate $\hat{\mathbf{R}}_s$, since the sums in $\hat{\mathbf{R}}_s$ involve also its "reciprocal" X_{hk}^{ij} , for which the collision kernel is not Maxwellian anymore:

$$\bar{B}_{hk}^{ij}(g) = \kappa_{ij}^{hk} \left(1 + \frac{\delta_{ij}^{hk}}{g^2} \right)^{1/2}$$

Monoatomic gases

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$$(\bar{B}^{sr}_{sr}(g)=\eta^{sr}_{sr}g,$$

Hard spheres
$$(\bar{B}^{sr}_{sr}(g)=\eta^{sr}_{sr}g, \qquad \Delta_{sr}=eta^{1/2}_{sr}|\mathbf{u}_s-\mathbf{u}_r|)$$

$$\hat{\mathbf{R}}_{sr} = -\frac{\eta_{sr}^{sr}\mu_{sr}N_{s}N_{r}}{\sqrt{\pi}\beta_{sr}^{1/2}}\left(\mathbf{u}_{s} - \mathbf{u}_{r}\right)\left[\frac{4\Delta_{sr}^{4} + 4\Delta_{sr}^{2} - 1}{4\Delta_{sr}^{3}}\sqrt{\pi}\operatorname{erf}\left(\Delta_{sr}\right) + \frac{2\Delta_{sr}^{2} + 1}{2\Delta_{sr}^{2}}\exp\left(-\Delta_{sr}^{2}\right)\right]$$

$$\begin{split} \hat{S}_{sr} &= -\mu_{sr} N_s N_r \eta_{sr}^{sr} \beta_{sr}^{-1/2} \left\{ \frac{1}{\alpha_{sr}} \left[\left(\Delta_{sr}^2 + 1 - \frac{1}{4\Delta_{sr}^2} \right) \frac{\text{erf}(\Delta_{sr})}{\Delta_{sr}} \right. \right. \\ &+ \left(1 + \frac{1}{2\Delta_{sr}^2} \right) \frac{\text{e}^{-\Delta_{sr}^2}}{\sqrt{\pi}} \right] \left(\frac{m_s}{2KT_s} \mathbf{u}_s + \frac{m_r}{2KT_r} \mathbf{u}_r \right) \cdot \left(\mathbf{u}_s - \mathbf{u}_r \right) + \frac{2K(T_s - T_r)}{m_s + m_r} \\ &\times \left[\left(\Delta_{sr}^4 + 3\Delta_{sr}^2 + \frac{3}{4} \right) \frac{\text{erf}(\Delta_{sr})}{\Delta_{sr}} + \left(\Delta_{sr}^2 + \frac{5}{2} \right) \frac{\text{e}^{-\Delta_{sr}^2}}{\sqrt{\pi}} \right] \right\} \end{split}$$

Chemically reacting mixture

Assumptions:

Mixture of 4 gases that, besides elastic collisions, are subject to a bimolecular and reversible chemical reaction

$$A_1 + A_2 \rightleftharpoons A_3 + A_4$$

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In equations for number densities, suitable reactive contributions appear:

$$Q_s^{\rm ch} = \lambda_s \, B_{\rm ch}^0 \left[\left(\frac{\mu_{12}}{\mu_{34}} \right)^{3/2} N_3 N_4 - N_1 N_2 \right], \qquad \lambda_1 = \lambda_2 = -\,\lambda_3 = -\,\lambda_4 = 1$$

(Analogous chemical contributions in equations for \mathbf{u}_s and T_s)

In space homogeneous conditions, seven independent first integrals are in order:

$$N_{s} + N_{r} = N_{s}^{0} + N_{r}^{0} \qquad (s, r) = (1, 3), (1, 4), (2, 4)$$

$$\mathbf{u} = \frac{1}{\rho^{0}} \sum_{s=1}^{4} \rho_{s} \mathbf{u}_{s} = \mathbf{u}^{0}$$

$$T = \frac{1}{3KN^{0}} \sum_{s=1}^{4} \rho_{s} u_{s}^{2} + \frac{1}{N^{0}} \sum_{s=1}^{4} N_{s} T_{s} = T^{0}$$
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(2)

⇒ The evolution actually takes place in a 13-dimensional subspace, once initial conditions are given, and independent variables may be chosen, for instance, as

$$N_1$$
, \mathbf{u}_2 , \mathbf{u}_3 , \mathbf{u}_4 , T_2 , T_3 , T_4

Collision equilibria

The "collision" operator in the Euler equations vanishes at the "physical" equilibrium

$$\mathbf{u}_1 = \mathbf{u}_2 = \mathbf{u}_3 = \mathbf{u}_4 (= \mathbf{u})$$
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$$\frac{N_1 N_2}{N_3 N_4} = \left(\frac{\mu_{12}}{\mu_{34}}\right)^{\frac{3}{2}} \equiv \xi$$

(a 7 parameter family with a single velocity and a single temperature, plus mass action law for densities)

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and then one gets for densities a quadratic equation with only one admissible (positive) solution

$$\begin{split} N_1^* &= \tfrac{1}{2(1-\xi)} \left\{ \left[\left(\xi(2N_1^0 + N_3^0 + N_4^0) - (N_1^0 - N_2^0) \right)^2 + 4\xi(1-\xi)(N_1^0 + N_3^0) \right. \\ & \left. \times \left(N_1^0 + N_4^0 \right) \right]^{1/2} - \left[\xi(2N_1^0 + N_3^0 + N_4^0) - (N_1^0 - N_2^0) \right] \right\}_{\text{M Bigi-Matrix}} \end{split}$$

M. Bisi – p. 22

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- Analogously for temperatures equations

Entropy dissipation

We consider the restriction of the classical reactive H-functional to the finite dimensional subspace of distribution functions defined by the fast collision equilibria $f_s^M(N_s, \mathbf{u}_s, T_s)$

$$\hat{H} = \sum_{s=1}^{4} N_s \left[\log \left(\frac{N_s}{m_s^3} \right) + \frac{3}{2} \log \left(\frac{m_s}{2\pi K T_s} \right) \right]$$

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It is possible to prove that it attains its minimum at the unique admissible equilibrium point and, in the space independent case, formal derivation yields

$$\partial_t \hat{H} = \sum_{s=1}^4 (\partial_t N_s) \left[\log \left(\frac{N_s}{m_s^3} \right) + \frac{3}{2} \log \left(\frac{m_s}{2\pi K T_s} \right) \right] - \frac{3}{2} \sum_{s=1}^4 N_s \frac{\partial_t T_s}{T_s} \le 0$$

Numerical examples

Reference case

Initial data and averaged collision frequencies

$$\begin{split} N_1^0 &= 2 \;, \quad N_2^0 = 4 \;, \quad N_3^0 = 3 \;, \quad N_4^0 = 1 \\ \mathbf{u}_1^0 &= (2,2,2) \;, \quad \mathbf{u}_2^0 = (4,4,4) \;, \quad \mathbf{u}_3^0 = (1,1,1) \;, \quad \mathbf{u}_4^0 = (3,3,3) \\ T_1^0 &= 2 \;, \quad T_2^0 = 3 \;, \quad T_3^0 = 4 \;, \quad T_4^0 = 1 \\ \bar{B}^{sr}_{sr} &= 1 \;, \quad B^0_{\mathsf{ch}} = 1 \;, \quad B^1_{\mathsf{ch}} = 0.5 \end{split}$$

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$$m_1 = 1.2$$
, $m_2 = 0.8$, $m_3 = 0.95$, $m_4 = 1.05$

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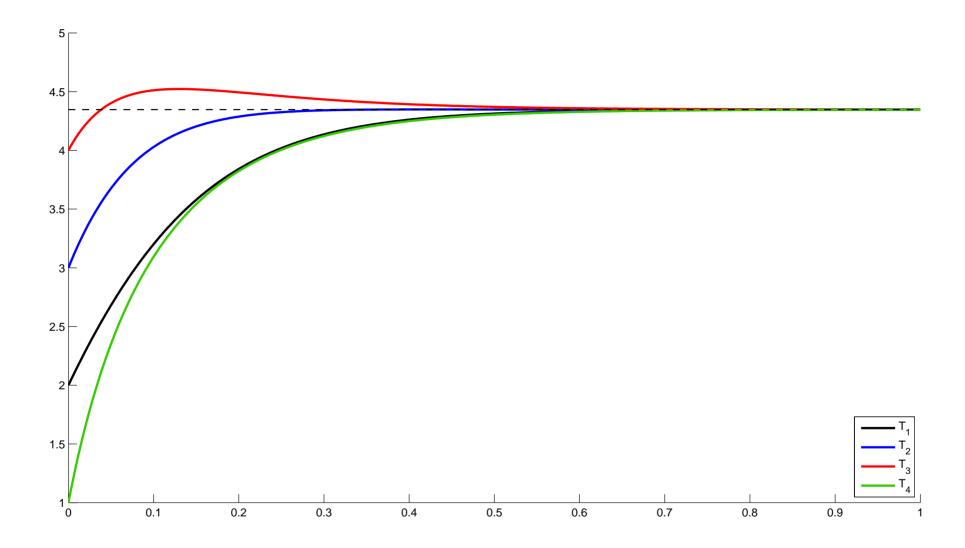
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Equilibrium values and relaxation rates are strongly depending on

$$\xi \equiv \left(\frac{\mu_{12}}{\mu_{34}}\right)^{\frac{3}{2}} = 0.94$$

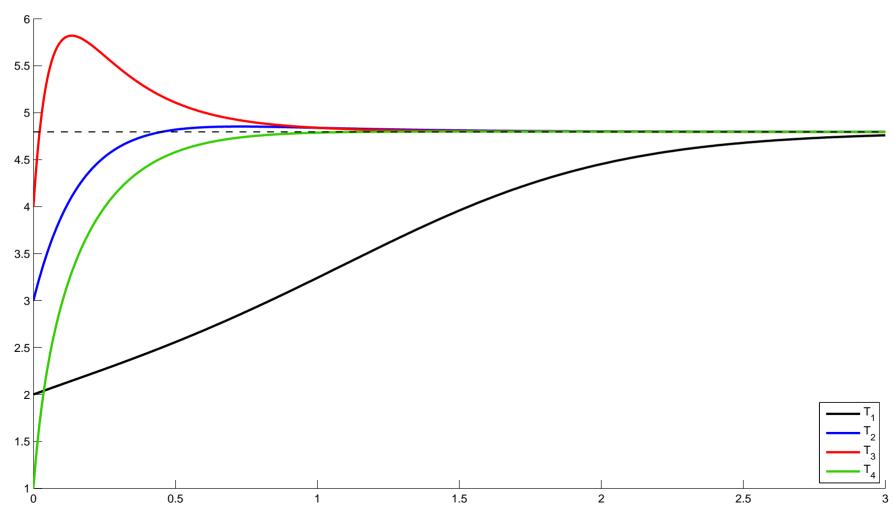
Temperatures evolution



Ex. 2

$$m_1 = 0.02$$
, $m_2 = 1.98$, $m_3 = 0.95$, $m_4 = 1.05$
 $\Rightarrow \qquad \xi \equiv \left(\frac{\mu_{12}}{\mu_{34}}\right)^{\frac{3}{2}} = 0.0079$

Temperatures



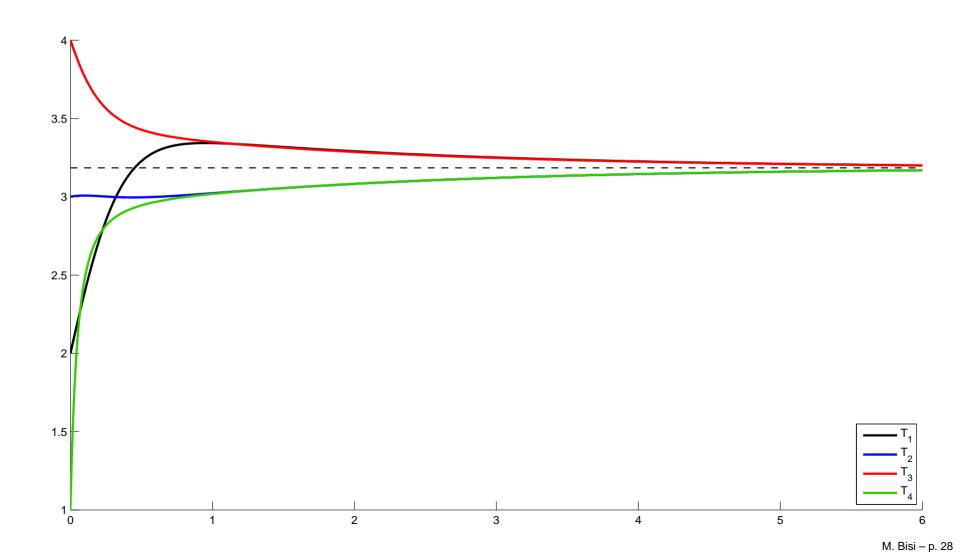
M. Bisi - p. 27

Ex. 3

$$m_1 = 0.02, \quad m_2 = 1.98, \quad m_3 = 0.05, \quad m_4 = 1.95$$

$$\Rightarrow \qquad \xi \equiv \left(\frac{\mu_{12}}{\mu_{34}}\right)^{\frac{3}{2}} = 0.26$$

Temperatures



comparison with results obtained from extended thermodynamics (T. Ruggeri, S. Simic)

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Thank you for your attention!